

COMPUTATION OF BINDING ENERGY OF BIOACTIVE CONSTITUENTS PRESENT IN CASSIA FISTULA SPECIES BY DFT APPROACH

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ABSTRACT

Bioactive compounds isolated from Cassia fistula plant were found to exhibit various pharmacological properties as revealed by the literature resources. In order to find the lead compounds among the eight pharmacological important constituents selected for our work, binding energy were computed by DFT approach using Gaussian software. B3LYP and HF method using three basis sets like STO-3G, 32-1G and 63-1G were adopted for the calculation of binding energy. Among the above constituents , Sennoside A was found to have good binding energy by both the methods. B3LYP method binding energies were found to be STO-3G (-3088.394), 32-1G(-3107.367) and 63-1G (-3124.647). Binding energy by STO-3G (-3066.066), 32-1G (-3088.914) and 63-1G (-3104.859) were observed by HF method. These results showed that among the bioactive constituents present in the cassia fistula plant, Sennoside A was found to have very good binding energy. Because of this very good binding energy, it may find as good medicine for the treatment of disease in future.

Keywords: Cassia fistula plant, DFT, B3LYP and HF.

No. of Tables: 2

No. of Figures: 8

No. of References:24

INTRODUCTION

Studies of phytochemical analysis of cassia fistula species have demonstrated anthrax quinone derivative and flavanoids present in the plant are widely used for its medicinal properties. Cassia fistula plant extract is used as an anti-periodic agent and in the treatment of rheumatism [1]. The leaf extract is found to have anti-tussive and wound healing properties [2]. In the Indian literature, this plant has been described to be useful against skin diseases, liver troubles, tuberculous glands and for the treatment of haematemesis, pruritus, leucoderma and diabetes [3-14]. The plant parts could be used as a therapeutic agent in the treatment of hypercholesterolaemia [15]. The plant has a high therapeutic value and can be used as anti-inflammatory, hypoglycemic, antipyretic, analgesic, anti-ulcer and laxative agent [16]. Leaves and flowers are both purgative like the pulp. Ashes from burnt pods mixed with little salt are used with honey taking 3-4 times to relieve cough. Flowers and pods are used as febrifugal, biliousness and astringent. The ethanolic extract of pods shows antifertility activity in female albino rats. The roots are used in fever, heart diseases, chest pain, joint pain, migraine, biliousness, and also applied in blood poisoning, anthrax, antidiarrhoeic, leprosy and antidiabetic, for the removal of abdominal obstruction and blood dysentery. The extract of the root lowered the blood sugar level up to 30% [17,18,19,20]. The bark possesses tonic and antidiarrhoeic properties, skin complaints, the powder or decoction of the bark is administered in leprosy,

jaundice, syphilis and heart diseases [21]. In our present work, binding energies of eight constituents have been computed using the B3LYP and HF methods by DFT approach.

MATERIALS & METHODS

The bioactive constituents present in the Cassia fistula plant were selected for our work as given below in figure. 1-8 [22].

Methods

DFT APPROACH

The structures of the constituents like Catechin, Fistic acid, Kaempferol, Rhein, Physcion, Rhein-8-glucoside, Procyanidin B2 and Sennoside A were drawn in Gauss View 5.0. The binding energies were calculated by B3LYP and HF methods using STO-3G, 3-21G, 6-31G basis sets [23,24&25]. The binding energy of the above compounds predicted by B3LYP and HF methods were given in **Table- 1 & Table- 2**

RESULT AND DISCUSSION

DFT calculation: B3LYP method

The binding energy calculated by STO-3G basis set were -1090.608, -941.280, -1015.450, -1014.247, -974.506, -1616.099, -2033.321 and -3083.394 a.u. for Catechin, Fistic acid, Kaempferol, Rhein, Physcion, Rhein-8-glucoside, Procyanidin B2 and Sennoside A compounds. Sennoside A was found to have good binding energy as -3083.394 a.u. Sennoside A has -3107.369 a.u. good binding energy compared to the other constituents binding energy, which were found to be -1099.005, -948.376, -1023.174, -1021.998, 984.376, -1628.638 and -2048.825 by 3-21G basis sets. The binding energies of the

above compounds calculated by 6-31G basis sets were found to be -1104.773,-953.367,-1028.548,-1027.374,-986.636,-1637.184 and -2059.539a.u..From the above data it was concluded that Sennoside A was found to have a good binding energy as -3123.647a.u. among the other bioactive constituents.

HF method

STO-3G

Among the bioactive compounds, Sennoside A have good binding energy as -3066.066a.u and the other constituents were found to be -1084.594,-935.926,-1009.783,-1008.594,-980.131,-1607.212 and -2021.731a.u.

3-21G

Sennoside A have good binding energy as -3088.914 compared to other bioactive constituents which were found to be -1092.623,-942.736,-1017.196,-1016.039,-987.468,-942.738,-1619.171 and -2036.566a.u.

6-31G

The binding energies observed using 6-31G basis sets for the

above constituents were found to be -1098.275,-947.636,-1022.466,-1021.317,-992.656,-953.374,-1627.543,-2036.566 and -3104.859a.u. From these data it was found that Sennoside A was found to be more stable among other compounds.

CONCLUSION

The binding energy of eight bioactive constituents were calculated by B3LYP and HF method using STO-3G, 3-21G and 6-31G. The B3LYP method revealed that Sennoside A was found to have good binding energy (-3123.647) and compared to other bioactive constituents -1104.773, -953.367, -1028.548, -1027.374, -986.636, -953.374, -1637.184 and -2059.539a.u. The HF methods of calculation of binding energies showed that Sennoside A was found to be good binding energy as -3104.859 compared to other bioactive compounds -1098.275,-947.636,-1022.466,-1021.317,-992.656,-953.374,-1627.543 and -2036.56a.u. The above computation of binding energy showed that among the eight bioactive compounds selected for the above work, Sennoside A was found to have a good binding energy.

Figures:

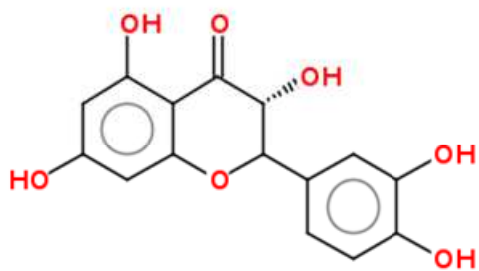


Figure. 1 Catechin

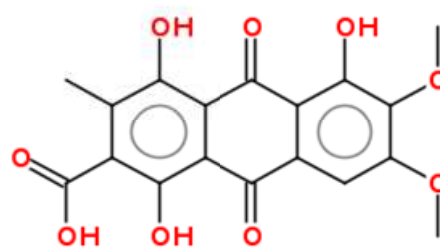


Figure. 2 Fisticacid

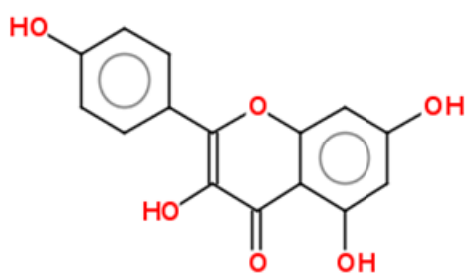


Figure. 3 Kaemferol

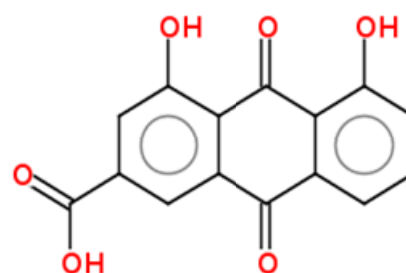


Figure. 4 Rhein

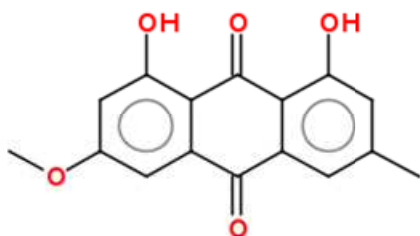


Figure. 5 Phycion

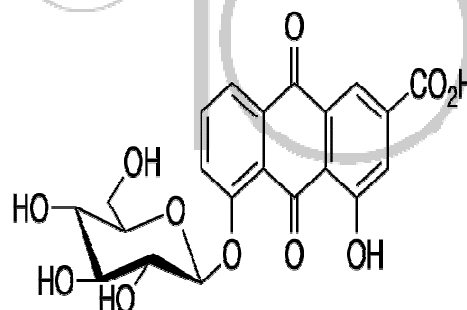


Figure. 6 Rhein-8-glucoside

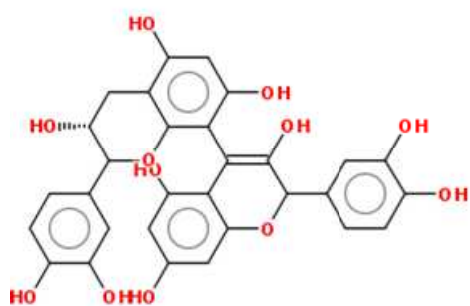


Figure. 7 Procyanidin B2

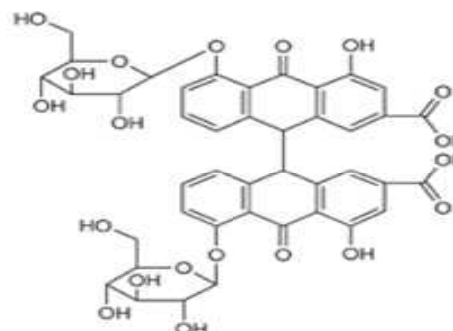


Figure. 8 Sennoside A

Table 1: Binding energy of compounds in B3LYP method

S.No	Compounds name	Basis sets		
		STO-3G	3-21G	6-31G
1	Catechin	-1090.608	-1099.005	-1104.773
2	Fistulic acid	-941.280	-948.376	-953.367
3	Kaempferol	-1015.450	-1023.174	-1028.548
4	Rhein	-1014.247	-1021.998	-1027.374
5	Physcion	-974.506	-984.376	-986.636
6	Rhein-8-glucoside	-1616.099	-1628.638	-1637.184
7	Procyanidin B2	-2033.321	-2048.825	-2059.539
8	Senoside A	-3083.394	-3107.369	-3123.647

Table 2: Binding energy of compounds in HF method

S.No	Compounds Name	Basis sets		
		STO-3G	3-21G	6-31G
1	Catechin	-1084.594	-1092.623	-1098.275
2	Fistulic acid	-935.926	-942.736	-947.636
3	Kaempferol	-1009.783	-1017.196	-1022.466
4	Rhein	-1008.594	-1016.039	-1021.317
5	Physcion	-980.131	-987.468	-992.656
6	Rhein-8-glucoside	-1607.212	-1619.171	-1627.543
7	Procyanidin B2	-2021.731	-2036.566	-2036.566
8	Senoside A	-3066.066	-3088.914	-3104.859

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